Sl.No. M21608 **Course Code: 3200105** 

## VINAYAKA MISSION'S RESEARCH FOUNDATIONS, SALEM (Deemed to be University)

### Pharm.D DEGREE EXAMINATION - July 2019 First Year

#### PHARMACEUTICAL INORGANIC CHEMISTRY

Time: Three hours	Maximum: 70 marks
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#### I. Write essays on any **TWO** questions:

 $(2 \times 15 = 30)$ 

- 1. Elaborate the principle and procedures involved in the limit test for arsenic and lead.
- 2. a) Summarize the preparation of 0.1M perchloric acid solution. (5)
  - b) outline the types of solvents used in non aqueous titration. (10)
- 3. a) Describe the methods used to measure and detect radioactivity. 10) (5)
  - b) Summarize the monograph of I<sup>131</sup>.

## II. Write short answers on any **SIX** questions:

 $(6 \times 5 = 30)$ 

- 4. What is ceriometry? Explain its advantage over other oxidizing agents
- 5. Explain the theory of acid base indicators with suitable examples
- 6. Explain the role of fluorides as anticaries agents.
- 7. Discuss sources of impurities in pharmaceutical substances.
- 8. Give the preparation, medicinal uses and storage condition of oxygen and nitrous oxide.
- 9. What are antimicrobials? Write the preparation and assay of chlorinated lime.
- 10. Briefly explain saline cathartics.
- 11. Outline the principle and procedure involved in gravimetric analysis with any one pharmaceuticals.

# III. Write short notes on any **FIVE** question:

 $(5 \times 2 = 10)$ 

- 12. Define ligands and give examples.
- 13. Write the role of chloride ion as electrolyte.
- 14. Recall the chemical reactions for assay of ammonium chloride by modified volhard's method.
- 15. What are essential and trace ions? Give examples
- 16. Write the uses of a) Zinc stearate b) Potassium citrate.
- 17. Define acidifiers with examples.